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Harris QCA 8e-Ch 00 1 of 15 1 A chemical firm was hired to monitor a nearby lake for possible mercury contamination Several samples were taken at different locations around the lake, including water samples, lake bottom samples, and plant

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Harris, Quantitative Chemical Analysis, 8e © 2011 W H Freeman [HL]n.org aq Quantitative Chemical Analysis, 8e W H Freeman Harris, © TABLE 22-2

Chem 274 Spring 2014 Principles of Analytical Chemistry

Chem 274 - Spring 2014 Principles of Analytical Chemistry Instructor: Professor John Head (Office: Bilger 236 - Email: johnh@hawaii.edu) Textbook: "Quantitative Chemical Analysis", Daniel C Harris, 8th edition, Freeman (2010) and "Solutions Manual" Develop an understanding of the physical principles of analytical chemistry

9. Analysis of an Acid-Base Titration Curve: The Gran Plot

9 Analysis of an Acid-Base Titration Curve: The Gran Plot Quantitative Chemical Analysis 44 The beauty of a Gran plot is that it enables us to use data taken before the end point to find the end point

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Gravimetric methods of analysis are in general the most accurate of the quantitative methods This is because mass can be measured with a high degree of precision and - + 8e- 3H 2O + NH 4 + D C Harris Quantitative Chemical Analysis 8th Ed, W H Freeman and Company, New York, Chapter 17 ...

Basic Statistics and error propagation

Critical values of TABLE 4-5 for rejection of outlier G Number of observations 10 11 12 15 20 (95% confidence) 1672 1822 1938 2032 2110 2176

HPLC - Instrumental Analysis

Regulated helium source Sol vent reservoirs Output check valve Inlet To waste Drain valve Pulse damper Inlet check valve Pump Sparger filter Solvent proportioning valve

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